# Morphology, physicochemical properties and antioxidant capacity of bee pollens

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**Abstract**: Six supposedly unifloral bee pollens of various botanical origins were characterised by morphometry, SEM, CIE  $L^*a^*b^*$  colour parameters and FTIR spectroscopy. Botanical origin and homogeneity of bee pollens were verified by colour and morphology of pollen grains. Water activity, moisture and antioxidant capacity of bee pollens were also evaluated. The results were discussed in terms of connection between botanical origin, composition and antioxidant properties of pollen materials.

Keywords: beekeeping products; colour; FTIR; SEM

Bee pollen is an apicultural product often used as food supplement. Its botanical origin is important as the main factor determining the presence of nutritional compounds and antioxidants (Almeida-Muradian et al. 2005; Human & Nicolson 2006; SILVA et al. 2006) responsible for various biological activities (Pascoal et al. 2014; Komosinska-Vassev et al. 2015; Denisow & Denisow-Pietrzyk 2016). This beekeeping product is interesting as natural food supplement that may supply human organism with a wide spectrum of health promoting nutrients (Kroyer & Hegedus 2001). However, it can be dangerous if consumed by some allergic people (MARTIN-MUNOZ et al. 2010). Chemical composition of bee pollen, including the presence of healthy and hazardous substances, depends on botanical origin of pollen grains, geographical locality, season and

storage conditions (GASPAROTTO SATTLER et al. 2015). Therefore, bee pollen authenticity is important for quality estimation. Bee pollen amino acids (SZCZĘSNA 2006), fatty acids (MANNING 2001; MARKOWICZ BASTOS et al. 2004; YOU et al. 2007), flavonoids (TOMÁS-BARBERÁN et al. 1989; CAMPOS et al. 1998), triterpene alcohols (XU et al. 2012) and carbohydrates (QIAN et al. 2008) can be used as biochemical markers of botanical origin. Otherwise, morphological signs of pollen grains, characteristic IR bands or colour parameters could be applied for bee pollen characterisation. Partially, FTIR spectroscopy has been used for estimation of bee pollen's composition (Anjos et al. 2017).

This article is devoted to characterisation of six supposedly unifloral bee pollens of various botanical origins by morphometry, scanning electron micros-

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copy (SEM), colour estimation and Fourier transform infrared (FTIR) spectroscopy. Water activity, moisture and antioxidant capacity of bee pollens were also evaluated.

### MATERIAL AND METHODS

Bee pollens. Six supposedly unifloral bee pollen samples (Table 1) were collected at different parts of Slovak Republic. Bee pollen catcher was used for this purpose under control of beekeepers. Homogenates of bee pollen were prepared for FTIR and colour measurements; intact loads were used for morphometry and SEM image analysis.

Morphology of bee pollen loads. Bee pollen loads were weighted individually on the Bosch SAE200 balance. Images of bee pollen loads were obtained by automatic macro magnifying glass Zeiss Discovery V12 (Carl Zeiss AG, Germany) using Zeiss AxioVision 4.7.1 software (module Automatic Measurement). The linear parameters, i.e. height (h) (mm) and width (w) (mm), were calculated from images of 100 randomly chosen loads of each bee pollen by the same software.

SEM images. Digital images (resolution up to  $3072 \times 2304 \, \mathrm{px}$ ) of bee pollen loads were obtained by scanning electron microscope Zeiss Evo LS 15 with variable pressure capability at 20 kV using a variable pressure secondary electron (VPSE) detector and Zeiss Axio Vision 4.7.1/LE 4.8.2 software (Carl Zeiss AG, Germany). Magnifications of  $\sim 200 \times$ ,  $\sim 500 \times$  and  $\sim 2000 \times$  were used to obtain images of load surface, ensembles of pollen grains and individual pollen grain, respectively. Polar (*P*) and equatorial (*E*) axes (µm) of individual pollen grains in the SEM images were measured using the same software (module Heasure).

**Colour estimation.** The CIE  $L^*a^*b^*$  parameters of bee pollen homogenates were calculated from

diffuse reflectance visible (Vis) spectra. Ten Vis spectra of each sample (380–780 nm, step 10 nm) were recorded on UV-Vis spectrophotometer UV4 (UNICAM, UK) using Labsphere holder (Labshere, USA) and Chroma 2.0 (ThermoSpectronic, USA) software.

FTIR spectroscopy. FTIR spectra (4000–400 cm<sup>-1</sup>, 64 scans and spectral resolution 2 cm<sup>-1</sup>) of bee pollens were recorded in KBr tablets on Nicolet 6700 FT-IR spectrometer using Omnic 7.0 (Thermo Scientific, USA) software. The smoothed and baseline corrected spectra were exported in the CSV format into Origin 6.0 (Microcal Origin, USA) software for graphical outputs.

Moisture and water activity. Karl Fischer titration method was done on Volumetric Karl Fischer titrator AF8 (Thermo Orion, USA) for determination of moisture in thawed fresh-frosted bee pollen tempered at 25°C. Water activity of bee pollens was measured using a LabMaster-aw neo (Novasina AG, Switzerland) and represented as the ratio of partial pressure of water vapour above the surface of the product to pressure of saturated water vapour at 25°C.

Antioxidant capacity. Two methods were used for evaluation of bee pollen antioxidant activity: 2,2-diphenyl-1-picrylhydrazyl (DPPH\*) and 2-2'-azino-di-[3-ethylbenzthiazoline sulfonate] (ABTS) photometric assays. The former one is based on decolouration of stable free radical DPPH' having a strong purple colour (BRAND-WILLIAMS et al. 1995); the latter one quantifies decolouration of blue-green coloured radical-cation ABTS\* generating from ABTS by metmyoglobin/ H2O2 (MILLER & RICE-EVANS 1996). ABTS assay was applied for the methanolic extracts from bee pollens; DPPH assay for both methanolic and aqueous extracts. Bee pollens (0.1 g) were mixed with 25 ml of water or methanol, exposed to ultrasound during 1 h and then filtrated using syringe filters. Commercial set Randox (Randox Laboratories, USA) was used for

Table 1. Specification of bee pollen samples

Sample	Specie	Collecting place <sup>a</sup>
1	oilseed rape (Brassica napus L. var. napus)	Slepčany
2	sunflower (Helianthus annuus L.)	Šahy
3	opium poppy (Papaver somniferum L.)	Dvory nad Žitavou
4	blue tansy (Phacelia tanacetifolia L.)	Zemianske Podhradie
5	black locust (Robinia pseudoacacia L.)	Mošovce
6	white clover (Trifolium repens L.)	Tesárske Mlyňany

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ABTS assay. The absorbance was measured at 517 nm (DPPH\*) and at 600 nm (ABTS\*\*). The resulting antioxidant capacity was expressed as an equivalent of TROLOX concentration (mmol/l) for trapping of the same amount of ABTS or as % of DPPH bleaching.

**Statistical evaluation**. For the morphological and CIE  $L^*a^*b^*$  parameters, moisture, water activity and antioxidant capacity, the results were compared among the samples using the ANOVA technique with Statistica 10 (StatSoft, USA). ANOVA indicated differences among the means, and Tukey (HSD) analysis of the differences was used for the comparisons between individual bee pollens. All data are reported as mean values with confidence intervals (P < 0.05).

#### RESULTS AND DISCUSSION

*Morphology of bee pollen loads*. Typical bee pollen loads of samples 1–6 are shown in Figure 1. There are evident differences in size, shape, surface micro-texture and colour of these loads. Variability of the mass, height



Figure 1. Typical bee pollen loads of samples 1-6

and width for bee pollen loads is represented in Table 2. The average masses of loads were in the range of ~8.4–15.3 g; variability coefficients confirmed moderate to very high variability in mass. The average heights and widths of loads were in the range of ~2.7–3.5 mm and ~3.3–3.8 mm, respectively; the corresponding variability coefficients confirmed moderate to high variability in these parameters. Bee pollen samples 3 and 4 consisted of loads with maximal and minimal values of all morphometric parameters, respectively. Surface texture of bee pollen loads is informative about palinological composition of bee pollen (Nôžκová *et al.* 2010). The differences in the surface texture of bee pollen loads were caused by specific structure of exine typical for mentioned botanical species.

Morphology of pollen grains. SEM images of pollen loads of 1-6 at three levels of magnification are shown in Figure 2. The samples 1, 2, 4 and 5 were pure unifloral bee pollens of botanic species mentioned above, whereas samples 3 and 6 were evidently contaminated by sunflower pollen grains. This admixture could appear during pollen collection by bee visiting flowers of more than one species. Otherwise, the surface scratches between bee pollen loads of different palinological origin may occur during bee pollen storage. Variability of pollen grains in bee pollen samples is summarised in Table 2. The grains of studied botanical species demonstrated significant difference in linear parameters, i.e. lengths at polar (P) and equatorial (E) axes. The shape index (SI, P/E ratio), which is a measure of elongation or roundness of pollen grains, also showed significant variability. Blue tansy pollen 4 is oblong (SI =  $2.03 \pm 0.20$ ), while sunflower pollen 2 is about roundish (SI =  $1.17 \pm 0.08$ ).

**Colour.** The CIE  $L^*a^*b^*$  parameters of bee pollen samples are summarised in Table 3. Among all bee

Table 2. Variability of bee pollen loads and grains in samples  $1\!-\!6$ 

Sample	Bee pollen loads (n = 100)		Pollen grains				
	m (mg)	h (mm)	w (mm)	N	P (µm)	<i>E</i> (μm)	SI
1	$11.47 \pm 0.18^{d}$	$3.31 \pm 0.018^{cd}$	$3.77 \pm 0.016^{b}$	40	$39.99 \pm 0.03^{d}$	$25.22 \pm 0.07^{c}$	1.59 ± 0.006
2	$10.70 \pm 0.19^{c}$	$3.38 \pm 0.008^{d}$	$3.72 \pm 0.008^{b}$	54	$45.27 \pm 0.01^{\rm f}$	$38.81 \pm 0.08^{e}$	$1.17 \pm 0.003$
3	$15.32 \pm 0.23^{\rm f}$	$3.45 \pm 0.004^{d}$	$3.77 \pm 0.006^{b}$	72	$37.67 \pm 0.01^{c}$	$25.90 \pm 0.03^{cd}$	$1.46 \pm 0.002$
4	$8.40 \pm 0.14^{a}$	$2.66 \pm 0.006^{a}$	$3.34 \pm 0.006^a$	105	$32.53 \pm 0.01^{a}$	$16.13 \pm 0.02^{a}$	$2.03 \pm 0.004$
5	$9.21 \pm 0.22^{b}$	$2.91 \pm 0.008^{b}$	$3.42 \pm 0.006^{a}$	87	$41.20 \pm 0.01^{\rm e}$	$21.56 \pm 0.03^{b}$	$1.92 \pm 0.002$
6	12.26 ± 0.20 <sup>e</sup>	$3.36 \pm 0.006^{d}$	$3.74 \pm 0.006^{b}$	90	$34.62 \pm 0.01^{b}$	$27.23 \pm 0.03^{d}$	1.27 ± 0.002

m – mass; h – height; w – width; P – polar axes; E – equatorial axes; SI – shape index (P/E); different letters indicate significant differences between bee pollens (P < 0.05)

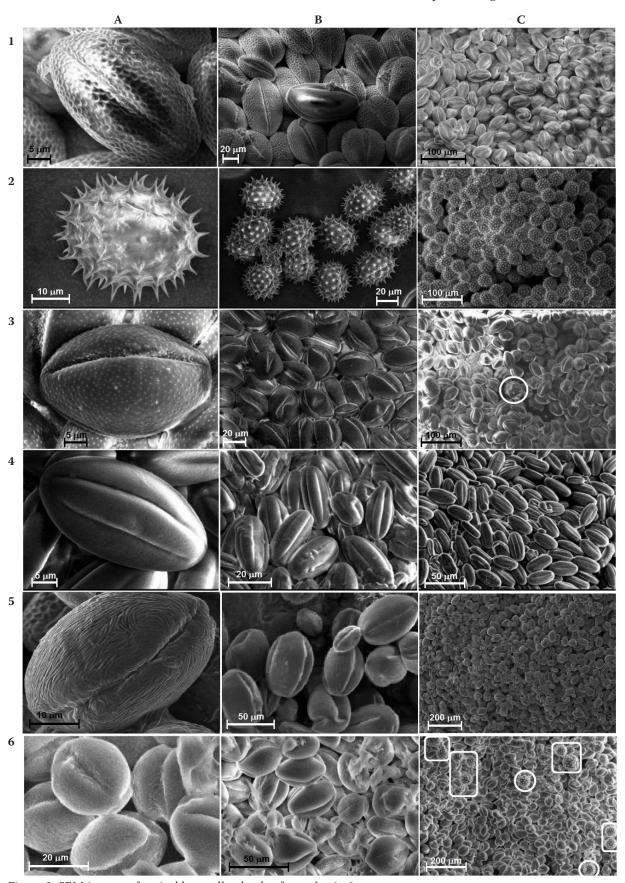


Figure 2. SEM images of typical bee pollen loads of samples  $1\!-\!6$ 

Table 3. CIE  $L^*a^*b^*$  parameters of bee pollens (n = 10)

Sample	$L^*$	a*	$b^*$	$C_{ m ab}^{\;*}$	$H_{\mathrm{ab}}{}^*$
1	$75.19 \pm 0.09^{d}$	$7.85 \pm 0.19^{b}$	29.86 ± 0.57 <sup>d</sup>	$30.88 \pm 0.54^{d}$	$1.313 \pm 0.007^{e}$
2	$57.95 \pm 0.38^{b}$	$19.99 \pm 0.41^{e}$	$25.84 \pm 1.50^{\circ}$	$32.69 \pm 1.26^{d}$	$0.911 \pm 0.027^{c}$
3	$51.40 \pm 0.18^{a}$	$14.79 \pm 0.58^{d}$	$8.99 \pm 1.35^{b}$	$17.42 \pm 0.53^{b}$	$0.543 \pm 0.074^{b}$
4	$51.75 \pm 0.27^{a}$	$1.11 \pm 0.22^{a}$	$-10.34 \pm 0.87^{a}$	$10.42 \pm 0.88^{a}$	$-1.467 \pm 0.018^{a}$
5	$75.23 \pm 0.17^{\rm d}$	$9.54 \pm 0.20^{\circ}$	$38.60 \pm 0.71^{e}$	$39.76 \pm 0.66^{e}$	$1.328 \pm 0.008^{e}$
6	$64.70 \pm 0.19^{c}$	$10.00 \pm 0.26^{c}$	$25.40 \pm 0.73^{c}$	$27.31 \pm 0.71^{c}$	$1.195 \pm 0.012^{d}$

Different letters indicate significant differences between bee pollens (P < 0.05)

pollens, samples 1 and 5 were the lightest ( $L^* \sim 75$ ) and samples 3 and 4 were the darkest ( $L^* \sim 51-52$ ). The colour parameter  $a^*$  (red–green axes) was positive for all bee pollens, and the colour parameters  $b^*$  (yellow–blue axes) and  $H_{ab}^*$  (hue) were also positive for all the samples except blue coloured bee pollen 4. For studied bee pollens the colour parameter  $C_{ab}^*$  (saturation) varied from  $\sim 10$  (sample 4) to  $\sim 40$  (sample 5). Differences in the CIE  $L^*a^*b^*$  parameters mentioned above are useful for distinguishing of bee pollens 1–6 according to their botanical origin.

Moisture and water activity. Pollen is known as highly hygroscopic material and thus susceptible for microbial or fungal growth (Carpes et al. 2009). Moisture is tightly connected with quality of bee pollen because wet conditions support grows of microorganisms, fungi and enzyme activities that may decrease safety and sensory properties of this product. The water activity indicates resistance of bee pollen to microbial contamination because if this value is too high, undesirable microorganisms may occur (Morgano et al. 2011). To avoid this, for better and longer storage, bee pollen is usually frozen or dried.

The fresh-frozen bee pollens 1-6 contained 11.5-19.5% of water. Water activity of these samples was in the range of 0.57-0.67 (Table 4). These values were

Table 4. Moisture and water activity of bee pollens (n = 5)

Sample	Moisture (% m/m)	Water activity
1	$14.784 \pm 0.195^{b}$	$0.665 \pm 0.036^{b}$
2	$15.524 \pm 0.808^{b}$	$0.594 \pm 0.056^{ab}$
3	19.539 ± 2.238°	$0.661 \pm 0.070^{ab}$
4	$15.741 \pm 2.211^{b}$	$0.574 \pm 0.032^{a}$
5	$11.534 \pm 0.743^{a}$	$0.595 \pm 0.047^{ab}$
6	$15.933 \pm 1.048^{b}$	$0.650 \pm 0.043^{ab}$

Different letters indicate significant differences between bee pollens (P < 0.05)

similar to those measured for fresh-frozen bee pollens and intermediate between those of freshly harvested and dried bee pollens, i.e. 0.52–0.59 (SAGONA et al. 2017), 0.62–0.82 (DE-MELO et al. 2015) and 0.21–0.55 (CARPES et al. 2009; ESTEVINHO et al. 2012; FEÁS et al. 2012; NOGUEIRA et al. 2012; DE-VEZA et al. 2015), respectively. All values obtained for thawed bee pollens were below 0.7 and thus low enough to avoid bacterial growth, but yeast growth is still possible (SAGONA et al. 2017).

FTIR spectra. Spectra of bee pollens 1–6 are shown in Figure 3. Broad band around 3300–3400 cm<sup>-1</sup> originated from OH stretching vibrations of water and hydroxyls. Two narrow bands near 2923 and 2852 cm<sup>-1</sup> were assigned to antisymmetric and symmetric CH<sub>2</sub> stretching vibrations in lipids, respectively (VLACHOS et al. 2006). Weak band or shoulder near 3012 cm<sup>-1</sup> arose from =CH stretching vibrations of unsaturated compounds. Several overlapped bands in the region of 1500–1800 cm<sup>-1</sup> mainly indicated the presence of carboxylic compounds, i.e. carbonic acids, esters and amides. Two intense bands near 1656 cm<sup>-1</sup> (amide I) and 1545 cm<sup>-1</sup> (amide II) confirmed the presence of proteins (Barth 2007). Less pronounced bands near 1740 and 1710 cm<sup>-1</sup> were assigned to

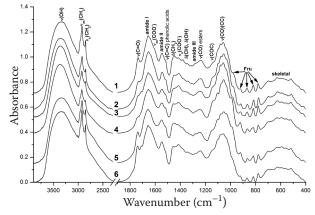


Figure 3. FTIR spectra of bee pollens 1-6

C=O stretching vibrations of ester and carboxylic groups, respectively. Weak band near 1518 cm<sup>-1</sup> arose from C=C stretching vibrations of phenolic acids (Kačuráková et al. 1999). The bands at 1451-1458 and 1378 cm<sup>-1</sup> were assigned to CH<sub>2</sub> and CH<sub>3</sub> bending vibrations in aliphatic groups. Band at 1414 cm<sup>-1</sup> arose from symmetric COO<sup>-</sup> stretching. The band or shoulder near 1280 cm<sup>-1</sup> was assigned to amide III vibrations in proteins. Broad band near 1240 cm<sup>-1</sup> originated from CO stretching vibrations in carboxyls and esters. Highly overlapped bands in the region of 900-1200 cm<sup>-1</sup> were assigned mainly to CO, CN a CC stretching vibrations of sugars and proteins. Bands at 777, 817, 865, 922, and 978 cm<sup>-1</sup> are typical for amorphous fructose (IBRAHIM et al. 2006). The bands below 700 cm<sup>-1</sup> could be assigned to various skeletal vibrations.

Antioxidant activity. Antioxidant activities of bee pollen samples are shown in Table 5. Both ABTS and DPPH methods confirmed that bee pollen samples had valuable antioxidant activity that is in agreement with literature (Leja et al. 2007; Morais et al. 2011; Feás et al. 2012). It is known that bee pollens contains significant amounts of antioxidants including hydrophilic (ascorbate, phenolic acids, flavonoids, etc.) and hydrophobic (unsaturated fatty acids, tocopherols and carotenoids) ones (RICE-EVANS et al. 1996; Krinsky 2001; Richard et al. 2008; RZEPECKA-STOJKO et al. 2015). According to DPPH assay, all methanolic extracts showed significantly higher antioxidant capacity than aqueous extracts due to the prevalence of mentioned antioxidants. Methanolic extracts from samples 1 and 4 showed maximal amounts of DPPH bleaching. In earlier investigation (FATRCOVÁ-ŠRAMKOVÁ et al. 2013), the antioxidant properties of unifloral bee pollens de-

Table 5. Antioxidant capacity of bee pollens 1-6

	ABTS $(n = 3)$	DPPH $(n = 5)$			
Sample	TROLOX (mm/l)	DPPH bleaching (%)			
		methanol	water		
1	$0.83 \pm 0.10^{a}$	$93.69 \pm 5.80^{d}$	$22.53 \pm 1.40^{a}$		
2	$1.90 \pm 0.20^{\rm b}$	$25.96 \pm 1.61^{a}$	$19.66 \pm 1.06^{a}$		
3	$2.08 \pm 0.25^{b}$	$61.30 \pm 3.84^{c}$	$43.95 \pm 2.63^{\circ}$		
4	$2.01 \pm 0.25^{b}$	$93.06 \pm 5.74^{d}$	$50.29 \pm 3.05^{\circ}$		
5	$2.00 \pm 0.22^{b}$	$92.10 \pm 5.65^{d}$	$47.21 \pm 2.83^{\circ}$		
6	$2.07 \pm 0.27^{\rm b}$	$43.55 \pm 2.58^{b}$	$34.28 \pm 2.16^{b}$		

Different letters indicate significant differences between bee pollens (P < 0.05)

creased in the order 1 > 3 > 2 that is in agreement with our results of DPPH assay obtained for methanolic extracts. By contrast, aqueous extracts from samples 1 and 2 demonstrated lower antioxidant capacity than those from the other samples according to both assays. Many reports (CAMPOS et al. 2003; LEJA et al. 2007; Pérez-Pérez et al. 2012) confirmed that high antioxidant capacity of bee pollen ethanolic or methanolic extracts correlated with high amount of polyphenols, mainly flavonoids and phenolic acids. The contribution of non-phenolic compounds to the whole antioxidant capacity of these extracts cannot be excluded. Moreover, free radical scavenging activity of bee pollen demonstrated significant decrease with aging that could be explained by instability of some antioxidants. CAMPOS et al. (2003) suggested that such decline in free radical scavenging can be used as a measure of bee pollen age if compare with fresh bee pollen of the same floral origin.

#### CONCLUSIONS

Six samples of unifloral bee pollen were characterised. Morphology and SEM images of bee pollen loads confirmed their botanical origin and homogeneity. Bee pollen samples demonstrated valuable antioxidant activity mainly due to the presence of methanol soluble phenolics and, to a lesser extent, carotenoids.

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